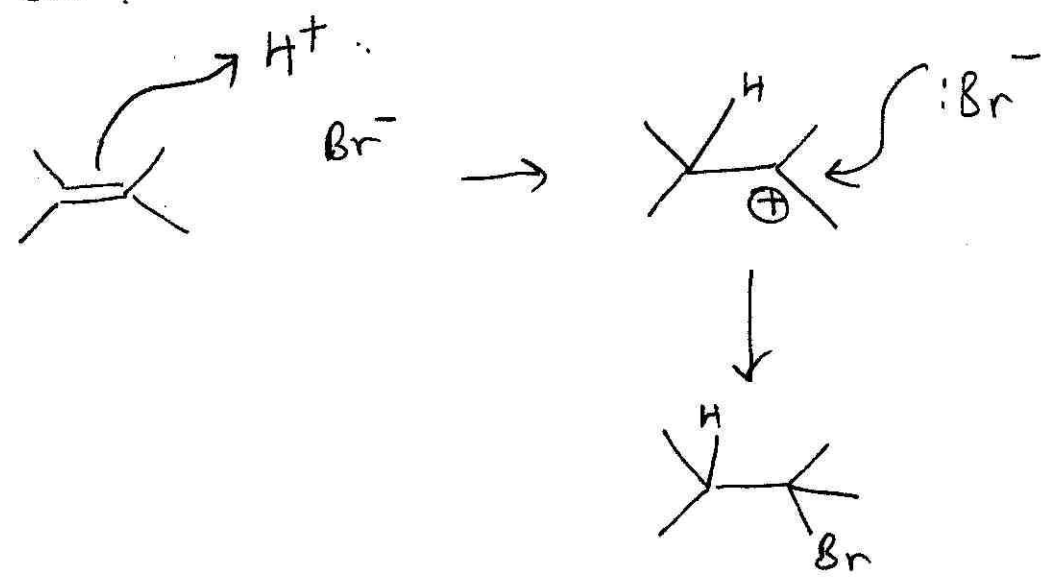


46. Electrophilic addition to the  $C=C$

46.1

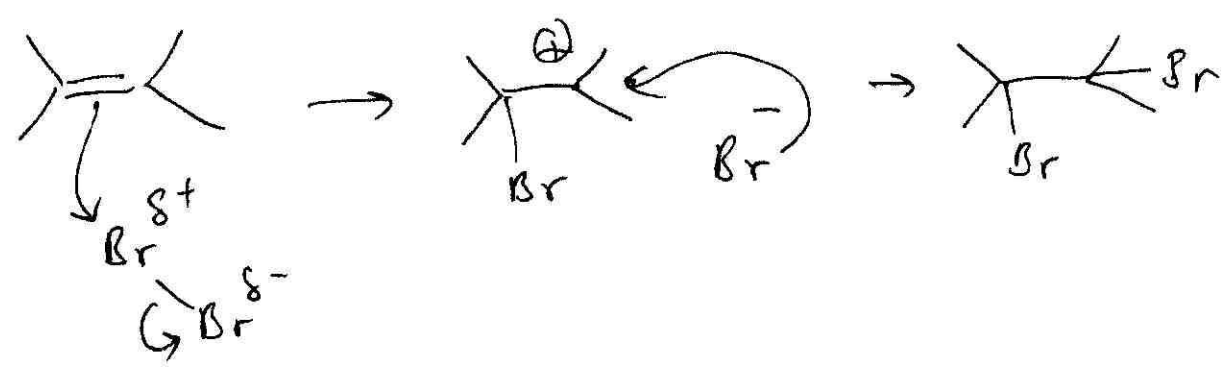
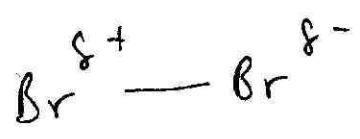
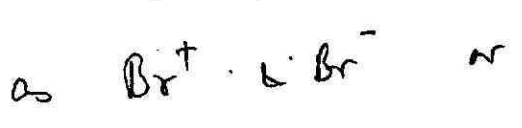
D) Another important reaction which you will study in organic chemistry is electrophilic addition to the double bond.



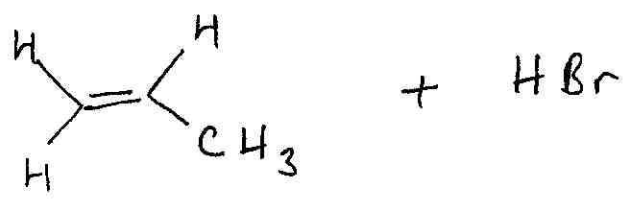
or another example:



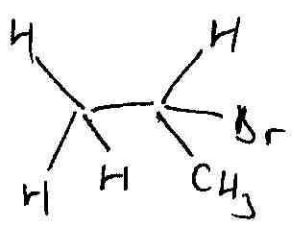
We can think of  $Br_2$



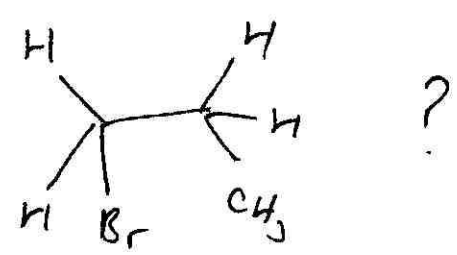
② Let's examine this reaction from an MO viewpoint & let's ask the question if we have



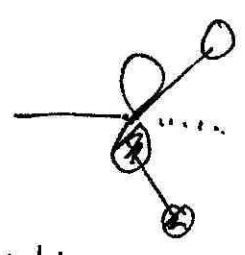
is the final product



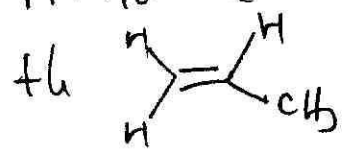
or



③ To answer this question we must decide whether -CH<sub>3</sub> is an electron donor or an electron withdrawer. As we discussed last time -CH<sub>3</sub> is an e<sup>-</sup> donor & its "key" orbital is

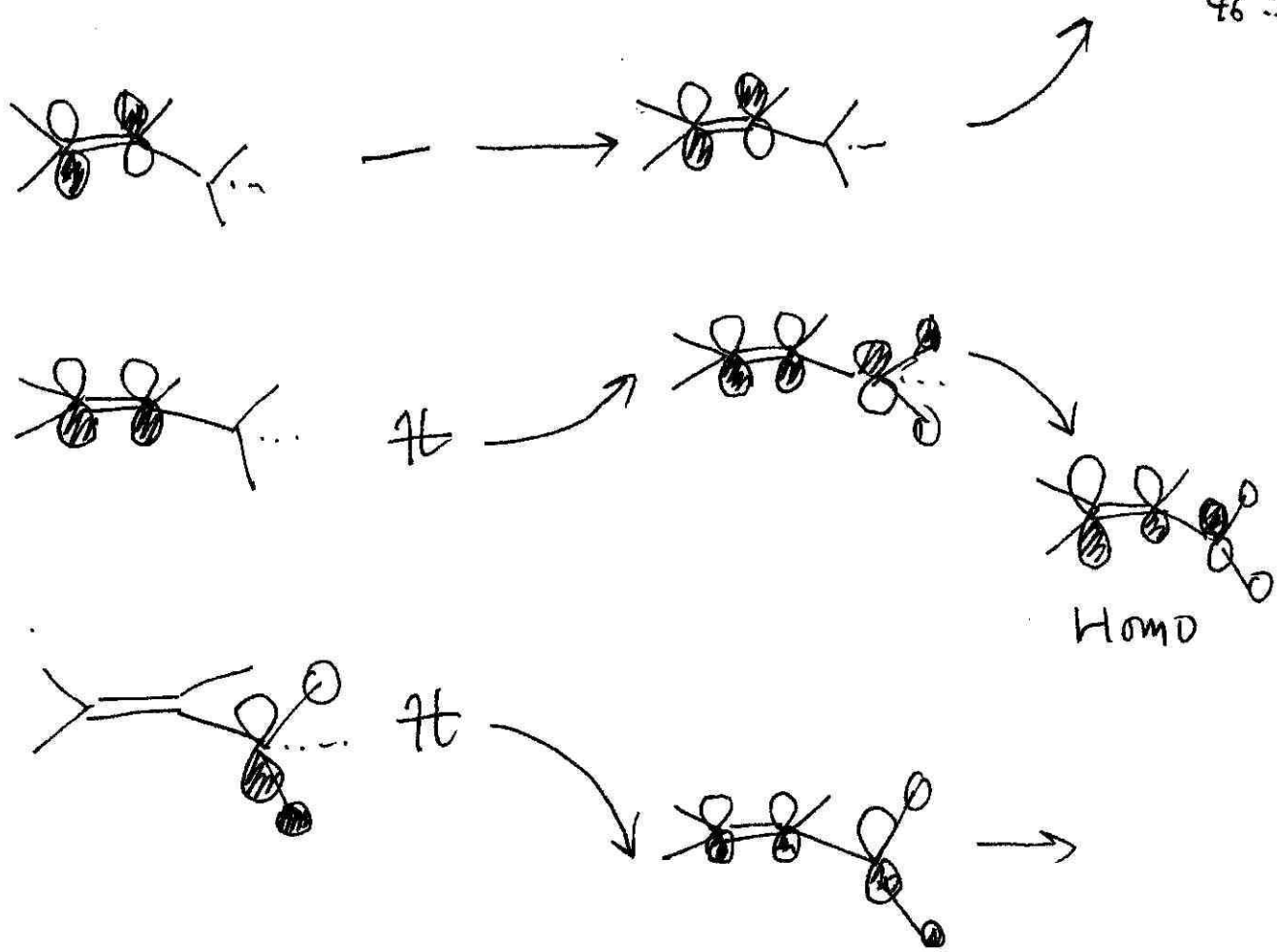


So let's see how this orbital charges the HOMO (note >C=C< is acting as nucleophile) of

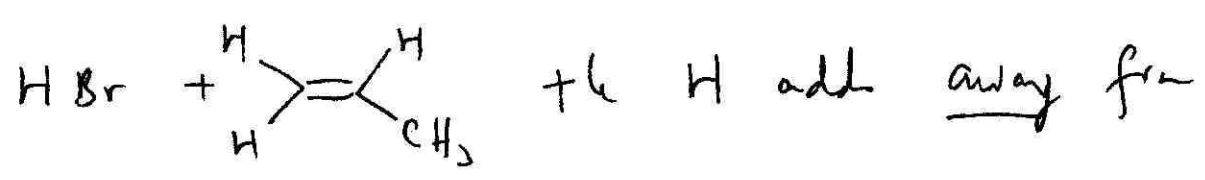


molecule.

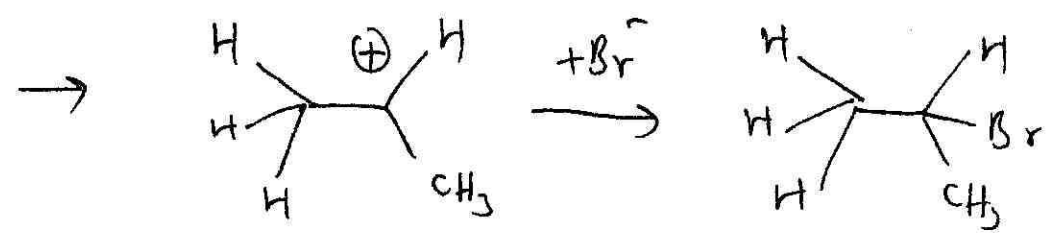
③



So in

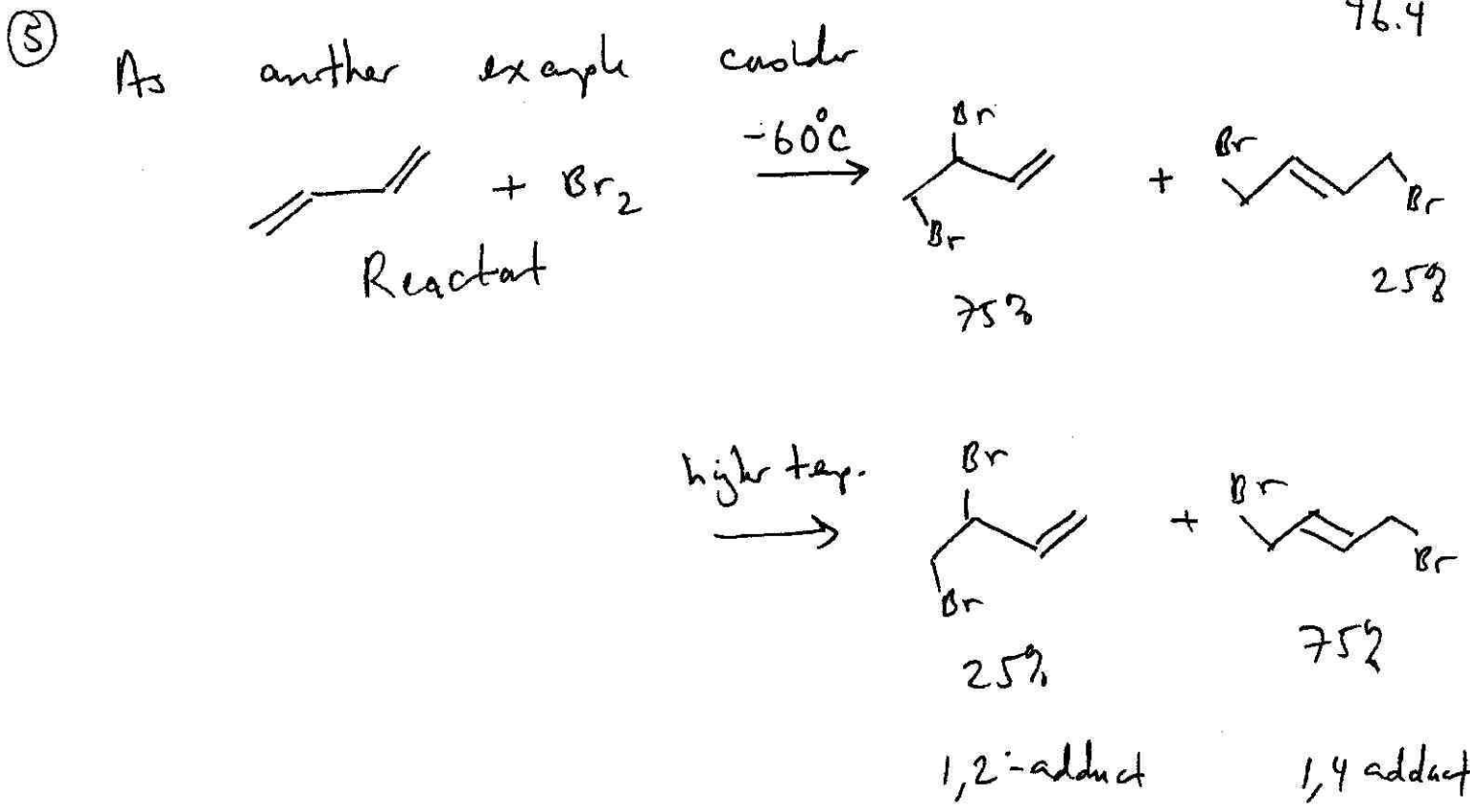


the  $\text{CH}_3$  (the  $e^-$  donor).

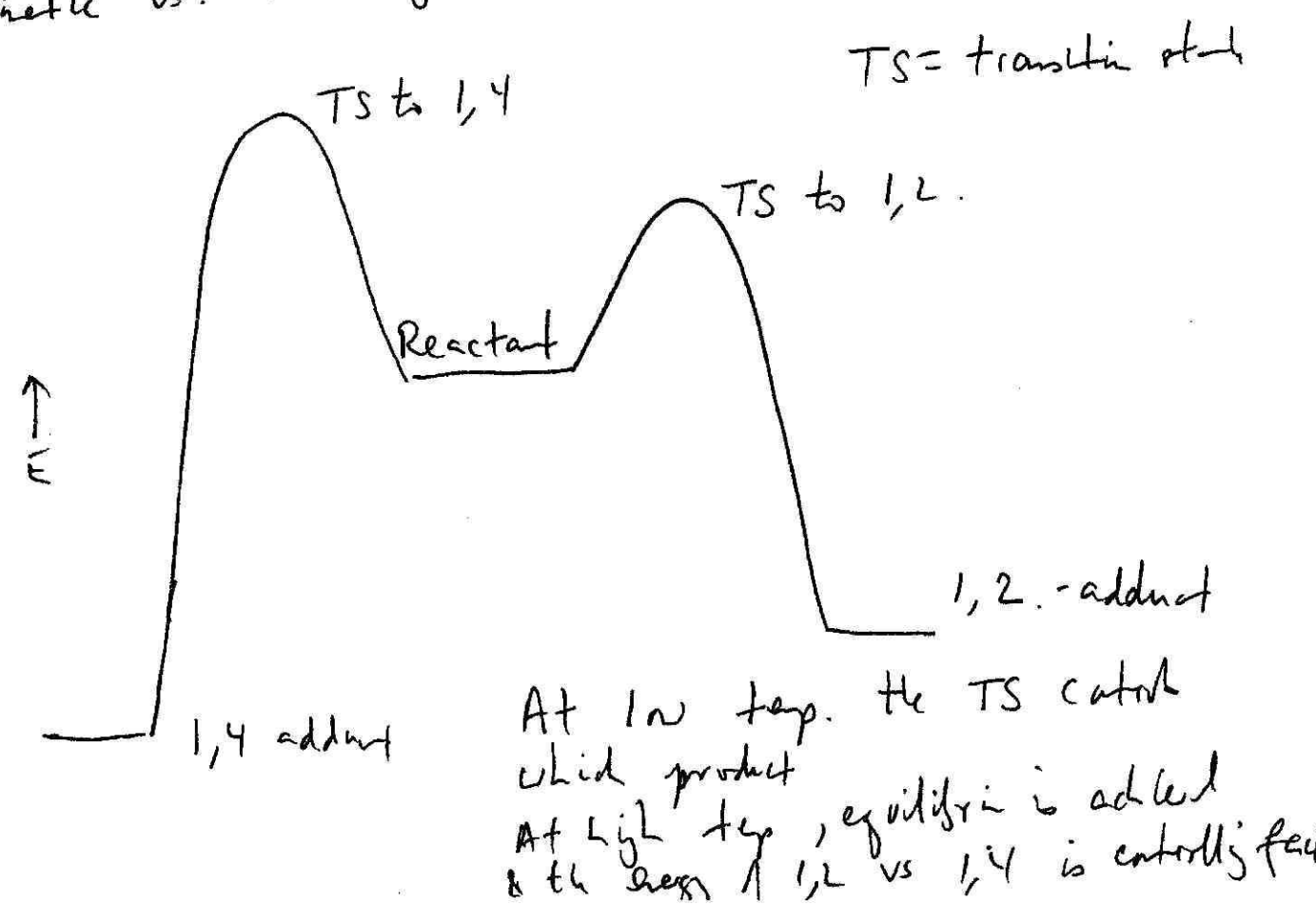


④

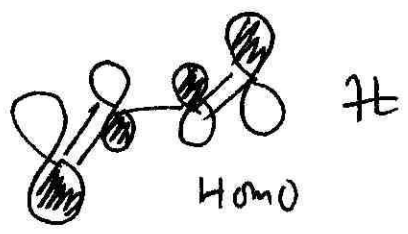
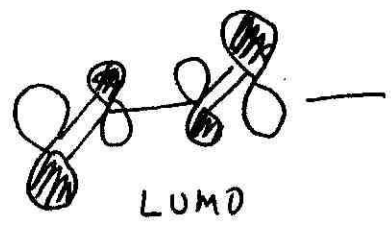
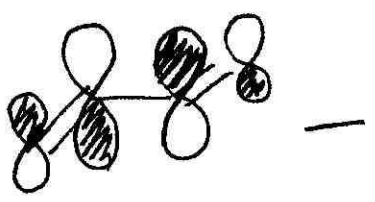
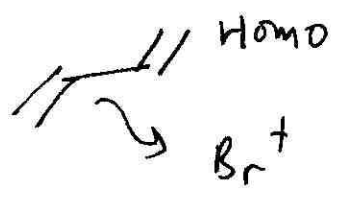
The halogen adds to the site with more  $\text{CH}_3$  groups: This is Markovnikov's rule.



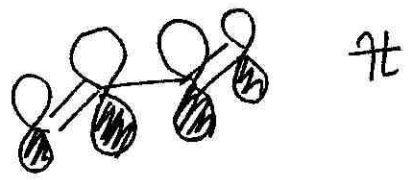
⑥ Chemists generally guess what this data means: kinetic vs. thermodynamic control.



⑨ In MO theory we will study what the kinetic product is. So let's follow the course of this reaction.



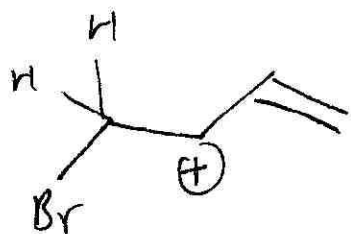
$\therefore$  1<sup>st</sup> Br addition to outer C.



8)

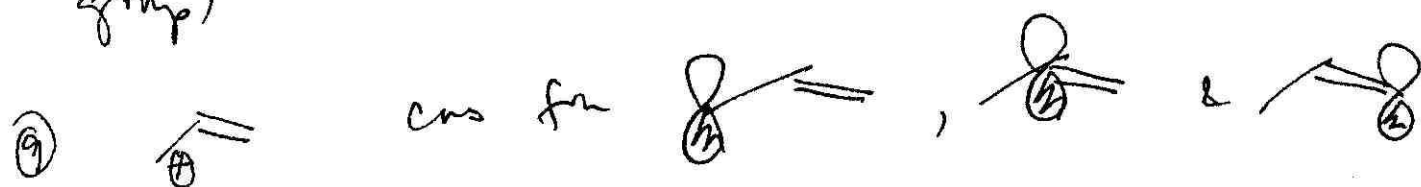
46.21

Now we have the molecule

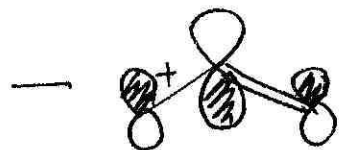


What does the MO diagram of this molecule look like? First let's draw the MO diagram of

$\text{CH}_2\text{Br}^-$  (which we will assume is an  $e^-$  donating group) & then we will consider the action of



MO diagram

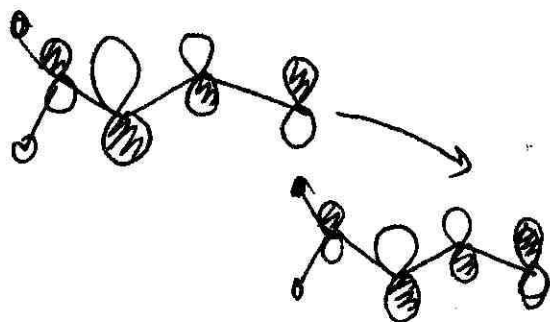


⑩ As we are interested in

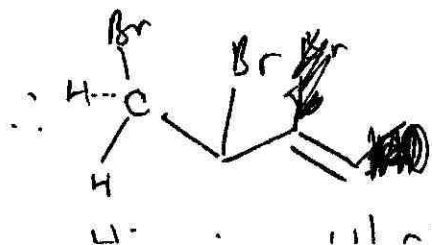
46.5



the organic piece is acting as the electrophile  
 we need to look at the LUMO.



↑  
 maximum overlap



will be the kinetic product.

what is observed.

**TABLE 15-3  
Principal Metal Toxins**

	Arsenic	Lead	Mercury	Thallium
Clinical tip-off	Red hands and burning feet with hyperhidrosis	Peripheral neuropathy that may appear to be a single nerve involvement (such as a wristdrop or footdrop)	Severe spontaneous arm and leg pain	Alopecia
Exposure	Homicide attempts; insecticides; medicinal arsenic; Paris green; accidental contamination; Fowler's solution (potassium arsenite)	Industrial ingestion; tetraethyl gasoline; lead paint; burning lead batteries; eating from pewter or dishware with a glaze containing lead; melting for purposes of molding	Ingestion of methyl mercury (Minamata disease), especially from fish in polluted areas; industrial exposure; antifungal treatment of grain	Homicide; insecticide; rodent poison
Clinical syndrome	"Stocking-glove," mainly sensory neuropathy; severe pain and paresthesias, especially of feet; "burning feet and hands," red hands with hyperhidrosis and subsequent motor neuropathy involving distal muscles of hands and feet	Primarily a motor neuropathy that frequently may appear as though single nerves are involved, such as the radial nerve (wristdrop), the median nerve (thenar atrophy), or the peroneal nerve (footdrop); painful joints; cerebral edema in children	Dementia with primary motor neuropathy; occasional sensory stocking-glove neuropathy; acrodynia (pink disease) in infants and young children	Distal sensorimotor neuropathy of stocking-glove type with alopecia
Diagnosis	24-hour urine analysis; hair analysis; blood arsenic level	Blood lead level; 24-hr urine analysis	24-hr urine analysis	24-hr urine analysis
Treatment	Penicillamine 250 mg 4 times a day (may also use BAL or EDTA)	Penicillamine 250 mg 4 times a day (may also use BAL or EDTA)	Penicillamine 250 mg 4 times a day (may also use BAL or EDTA)	Diphenylthio-carbazone or sodium dicarbamate





2A  
2

3A 4A 5A 6A 7A  
13 14 15 16 17



3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.01218										
11 <b>Na</b> Sodium 22.98977	12 <b>Mg</b> Magnesium 24.305	3B 3	4B 4	5B 5	6B 6	7B 7	8 8	9 9	10 10	1B 11	2B 12

19 <b>K</b> Potassium 39.0983	20 <b>Ca</b> Calcium 40.078	21 <b>Sc</b> Scandium 44.9559	22 <b>Ti</b> Titanium 47.88	23 <b>V</b> Vanadium 50.9415	24 <b>Cr</b> Chromium 51.9961	25 <b>Mn</b> Manganese 54.9380	26 <b>Fe</b> Iron 55.847	27 <b>Co</b> Cobalt 58.9332	28 <b>Ni</b> Nickel 58.6934	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.39
37 <b>Rb</b> Rubidium 85.4678	38 <b>Sr</b> Strontium 87.62	39 <b>Y</b> Yttrium 88.9059	40 <b>Zr</b> Zirconium 91.224	41 <b>Nb</b> Niobium 92.9064	42 <b>Mo</b> Molybdenum 95.94	43 <b>Tc</b> Technetium (98)	44 <b>Ru</b> Ruthenium 101.07	45 <b>Rh</b> Rhodium 102.9055	46 <b>Pd</b> Palladium 106.42	47 <b>Ag</b> Silver 107.8682	48 <b>Cd</b> Cadmium 112.411
55 <b>Cs</b> Cesium 132.9054	56 <b>Ba</b> Barium 137.327	57 <b>*La</b> Lanthanum 138.9055	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.9479	74 <b>W</b> Tungsten 183.85	75 <b>Re</b> Rhenium 186.207	76 <b>Os</b> Osmium 190.2	77 <b>Ir</b> Iridium 192.22	78 <b>Pt</b> Platinum 195.08	79 <b>Au</b> Gold 196.9665	80 <b>Hg</b> Mercury 200.59
87 <b>Fr</b> Francium (223)	88 <b>Ra</b> Radium 226.0254	89 <b>+Ac</b> Actinium 227.0278	104 <b>Rf</b> Rutherfordium (261)	105 <b>Db</b> Dubnium (262)	106 <b>Sg</b> Seaborgium (263)	107 <b>Bh</b> Bohrium (262)	108 <b>Hs</b> Hassium (265)	109 <b>Mt</b> Meitnerium (268)	110 110	111 111	112 112

13 <b>Al</b> Aluminum 26.98154	14 <b>Si</b> Silicon 28.0855	15 <b>P</b> Phosphorus 30.97376	16 <b>S</b> Sulfur 32.06	17 <b>Cl</b> Chlorine 35.45	18 <b>Ar</b> Argon 39.948
49 <b>In</b> Indium 114.82	50 <b>Sn</b> Tin 118.710	81 <b>Tl</b> Thallium 204.3833	82 <b>Pb</b> Lead 207.2	83 <b>Bi</b> Bismuth 208.9804	84 <b>Po</b> Polonium (209)

*Lanthanide Series											
58 <b>Ce</b> Cerium 140.115	59 <b>Pr</b> Praseodymium 140.9077	60 <b>Nd</b> Neodymium 144.24	61 <b>Pm</b> Promethium (145)	62 <b>Sm</b> Samarium 150.36	63 <b>Eu</b> Europium 151.965	64 <b>Gd</b> Gadolinium 157.25	65 <b>Tb</b> Terbium 158.9254	66 <b>Dy</b> Dysprosium 162.50	67 <b>Ho</b> Holmium 164.9303	68 <b>Er</b> Erbium 167.26	69 <b>Tm</b> Thulium 168.9342
90 <b>Th</b> Thorium 232.0381	91 <b>Pa</b> Protactinium 231.0359	92 <b>U</b> Uranium 238.0289	93 <b>Np</b> Neptunium 237.048	94 <b>Pu</b> Plutonium (244)	95 <b>Am</b> Americium (243)	96 <b>Cm</b> Curium (247)	97 <b>Bk</b> Berkelium (247)	98 <b>Cf</b> Californium (251)	99 <b>Es</b> Einsteinium (252)	100 <b>Fm</b> Fermium (257)	101 <b>Md</b> Mendelevium (258)
† Actinide Series											
70 <b>Yb</b> Ytterbium 173.04	71 <b>Lu</b> Lutetium 174.967	102 <b>No</b> Nobelium (259)	103 <b>Lr</b> Lawrencium (260)								

# Price/gram most expensive elements


21 <b>Sc</b> \$131	22 <b>Ti</b> Titanium 47.88	23 <b>V</b> Vanadium 50.9415	24 <b>Cr</b> Chromium 51.9961	25 <b>Mn</b> Manganese 54.9380	26 <b>Fe</b> Iron 55.847	27 <b>Co</b> Cobalt 58.9332	28 <b>Ni</b> Nickel 58.6934	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.39	31 <b>Ga</b> Gallium 69.723	
39 <b>Y</b> Yttrium 88.9059	40 <b>Zr</b> Zirconium 91.224	41 <b>Nb</b> Niobium 92.9064	42 <b>Mo</b> Molybdenum 95.94	43 <b>Tc</b> Technetium (98)	44 <b>Ru</b> \$2.50	45 <b>Rh</b> \$71	46 <b>Pd</b> \$7	47 <b>Ag</b> \$0.25	48 <b>Cd</b> Cadmium 112.411	49 <b>In</b> Indium 114.82	50 <b>Sn</b> Tin 118.710
57 <b>*La</b> Lanthanum 138.9055	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.9479	74 <b>W</b> Tungsten 183.85	75 <b>Re</b> \$12	76 <b>Os</b> \$95	77 <b>Ir</b> \$6	78 <b>Pt</b> \$31	79 <b>Au</b> \$15	80 <b>Hg</b> \$0.02	81 <b>Tl</b> Thallium 204.3833	82 <b>Pb</b> Lead 207.2

Sources: <http://www.minerals.usgs.gov/minerals>  
<http://metalsplace.com/metalsboard> (Os)



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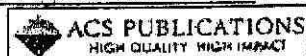
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1: [J Agric Food Chem.](#) 2000 Oct;48(10):4830-4.

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## Organoleptic impact of 2-methoxy-3-isobutylpyrazine on red bordeaux and loire wines. Effect of environmental conditions on concentrations in grapes during ripening.

Roujou de Boubée D, Van Leeuwen C, Dubourdieu D.

Faculte d'Oenologie, Universite Victor Segalen Bordeaux II, 351 Cours de la Liberation, 33405 Talence Cedex, France. droujou@u-bordeaux2.fr

The 2-methoxy-3-isobutylpyrazine content in grapes and red wines was assayed by stable isotope dilution gas chromatography-mass spectrometry, following vapor extraction and purification on a cation resin microcolumn. The threshold beyond which the green bell pepper character is marked in wines has been determined. From a comparison of the 2-methoxy-3-isobutylpyrazine concentrations of 50 red Bordeaux and Loire wines from different vintages and grape varieties (Cabernet Sauvignon, Cabernet franc, and Merlot) with the intensity of the green bell pepper character as perceived on tasting, the threshold value was estimated to be 15 ng/L. Statistical analysis of the 2-methoxy-3-isobutylpyrazine concentrations of 89 red Bordeaux wines showed that Cabernet wines were more commonly affected by this vegetative character. Changes in the 2-methoxy-3-isobutylpyrazine concentration as the grapes ripen are affected by the environmental and cultural conditions (soil, climate, training system, etc.). A very good correlation was shown between the breakdown of malic acid and 2-methoxy-3-isobutylpyrazine as the grapes ripened, irrespective of grape variety, type of soil, or weather conditions.

PMID: 11052741 [PubMed - indexed for MEDLINE]

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## Leffingwell & Associates

Odor Properties &amp; Molecular Visualization

# Pyrazines

## DETECTION THRESHOLDS & Molecular Structures

### 2-Methylpyrazine... $C_5H_6N_2$

Odor Detection Threshold (in water) = 60,000 ppb

Green, nutty, cocoa, musty, potato, fishy-ammoniacal notes

### 2-Ethylpyrazine... $C_6H_8N_2$

Odor Detection Threshold (in water) = 6,000 ppb

Musty, nutty, buttery, peanut odor; chocolate-peanut taste

### 2,3-Dimethylpyrazine... $C_6H_8N_2$

Odor Detection Threshold (in water) = 2,500 ppb

Green, nutty, potato, cocoa, coffee, caramel, meaty notes

### 2,5-Dimethylpyrazine... $C_6H_8N_2$

Odor Detection Threshold (in water) = 800 ppb

Chocolate, roasted nuts, earthy; chocolate taste

### 2,6-Dimethylpyrazine... $C_6H_8N_2$

Odor Detection Threshold (in water) = 200 ppb

Chocolate, roasted nuts, fried potato odor

### 2,3,5-Trimethylpyrazine... $C_7H_{10}N_2$

Odor Detection Threshold (in water) = 400 ppb

Nutty, baked potato, roasted peanut, cocoa, burnt notes

### 2,3,5,6-Tetramethylpyrazine... $C_8H_{12}N_2$

Odor Detection Threshold (in water) = 1000 ppb

Weak, nutty, musty, chocolate odor; chocolate taste

### 2-Ethyl-3-methylpyrazine... $C_7H_{10}N_2$

Flavor Detection Threshold (in water) = 0.4 ppb

Potato, burnt nutty, roasted, cereal, earthy

### 2-Ethyl-5-methylpyrazine... $C_7H_{10}N_2$

Odor Detection Threshold (in water) = 100 ppb

Nutty, roasted, somewhat "grassy"

### 2-Ethyl-3,5-dimethylpyrazine... $C_8H_{12}N_2$



A European breed of paprika

**Used plant part**

Berry fruits. Removal of seeds and veins results in a less pungent and more brightly coloured product.

**Plant family**

**Solanaceae** (nightshade family).

**Sensory quality**

Sweet and aromatic. Some qualities show no pungency at all, others are fairly hot

**Main constituents**

The pungent principle, capsaicin, is contained only in small amounts, as low as 0.001 to 0.005% in ?mild? and 0.1% in ?hot? cultivars (see chiles for details on capsaicin).

Apart from capsaicin, the taste of paprika is mostly due to essential oil (<1%; with long-chain aliphatic hydrocarbons, fatty acids and their methyl esters); paprika scent is mostly due to a range of alkylmethoxy-pyrazines (e.g., 3-isobutyl 2-methoxy pyrazine, ?earthy? flavour). Ripe paprika contain up to 6% sugar.